

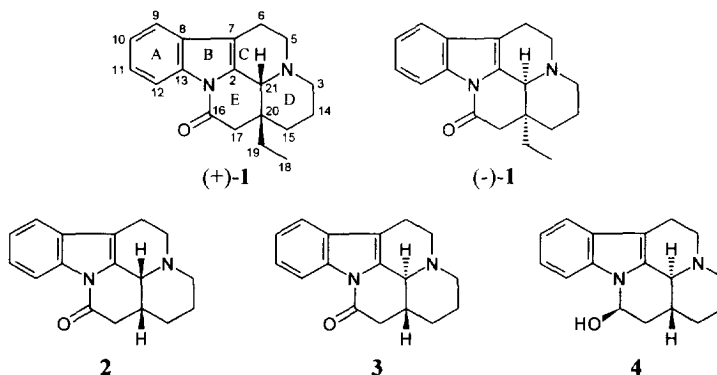
Acid-Catalysed Epimerization of 1-Substituted Indolo[2,3-*a*]quinolizidines: Stereoselective Routes to *cis*- and *trans*-Deethyleburnamonine Starting from the Same Ester Intermediate

Mauri Lounasmaa*, Lars Miikki, and Arto Tolvanen

*Laboratory for Organic and Bioorganic Chemistry,
Technical University of Helsinki, FIN-02150 Espoo, Finland*

Abstract: Efficient syntheses of the pharmacologically interesting deethyleburnamonines **2** and **3** are described. Independent routes to **2** and **3** start from the same, easily accessible ester **5**. The choice between the routes requires an acid-catalysed epimerization of **5**. Conformations of some intermediate indolo[2,3-*a*]quinolizidines are discussed. Copyright © 1996 Elsevier Science Ltd

Eburnamonine, as both its optically active forms (+)-**1** and (-)-**1**, is a highly valued drug of natural origin used in the treatment of cerebral disorders. In view of its wide use, this indole alkaloid of eburnamine-vincamine type is now a popular synthetic target.¹ Interesting pharmacological properties are also possessed by several derivatives of eburnamonine, including the two ethyl group lacking compounds *cis*-deethyleburnamonine (**2**) and *trans*-deethyleburnamonine (**3**).² These compounds are precursors of other important derivatives, of which vindeburnol (RU 24722) (**4**)³ is one of the most studied.

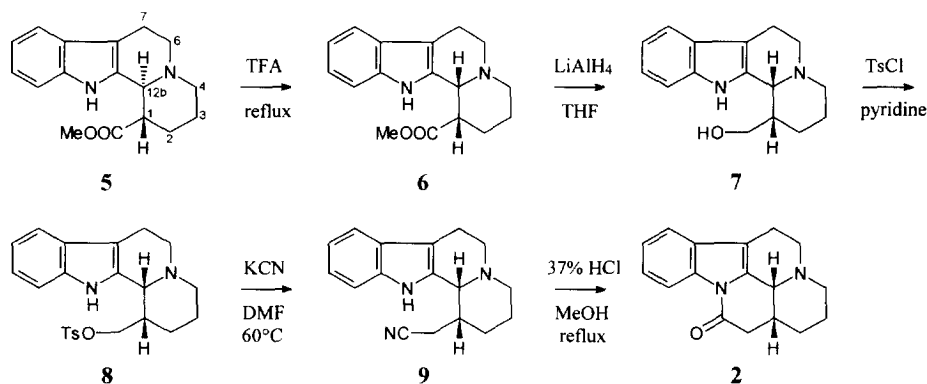


Much of the recent synthetic work carried out in our laboratory has been directed to alkaloids of eburnamine-vincamine type and their structurally close *Iboga* analogues, the tacamines.⁴ The interesting

epimerization behaviour of certain 1,3-disubstituted indolo[2,3-*a*]quinolizidines required in the synthesis of tacamine-type compounds^{4d,4c,4g} prompted us to examine some 1-substituted analogues. Instead of relying on the tedious oxidation/reduction procedures normally used with 1-substituted indolo[2,3-*a*]quinolizidines, we controlled the configurational relationship between H-1 and H-12b by a simple acid-catalysed epimerization, which led mainly to the isomer where the 1-substituent is axial (H-1 and H-12b *cis*). By choosing the proper starting compound, stereoselective routes to *cis*- and *trans*-deethyleburnamonine can be devised.

Results and Discussion

Ester **5**, which can be prepared in three steps from tryptophyl bromide and methyl nicotinate^{5,6}, was chosen as the starting compound for the synthesis of the deethyleburnamonines. Treatment of ester **5** with trifluoroacetic acid (TFA)⁷ under reflux for 16 h led to a mixture consisting of the starting compound and its isomer **6**, the latter of which was isolated from this mixture in 78% yield. The two esters were then converted to the two deethyleburnamonines *via* essentially identical homologation processes (Schemes 1 and 2). Reduction of **6** and **5** with LiAlH₄ gave in high yields the corresponding alcohols (**7** and **10**), which were converted to nitriles **9** and **12** *via* tosylates **8** and **11**, respectively. Cyclization of nitrile **9**⁸ in aq HCl/MeOH furnished (±)-*cis*-deethyleburnamonine (**2**) in 82% yield. Treatment of tosylate **11** with potassium cyanide in DMF yielded nitrile **12**, which exists partly in its tautomeric imine form **13** (Scheme 2). The ratio of **12** and **13** seems to be dependent on the pH, but the two compounds can be separated and characterized without difficulty. Mild acid treatment of a crude mixture of **12** and **13** afforded (±)-*trans*-deethyleburnamonine (**3**) from **11** in 55% yield (two steps).



The complete spectral data of ester **6** have not been published, although this compound has been mentioned in the literature.⁹ The ¹³C NMR data of **6** indicate that, in CDCl₃ solution, it does not exist purely as a C/D *trans*-fused indoloquinolizidine (conformation a), but there is also a considerable contribution of conformer c (C/D-*cis*) to the equilibrium.¹⁰ The same applies to nitrile **12**, which spectrally resembles the

corresponding 1-ethyl-1,2,3,4,6,7,12,12b-octahydroindolo[2,3-*a*]quinolizine.¹¹ Surprisingly, the ¹³C NMR analysis of alcohol **10** gave a different result. As Stoit and Pandit¹² have pointed out, in DMSO-*d*₆ there is an equilibrium between the two conformations as described above. However, we were able to run a spectrum also in CDCl₃, which showed that in this solvent (in very low concentration) alcohol **10** exists predominantly in conformation a (*C/D-trans*).

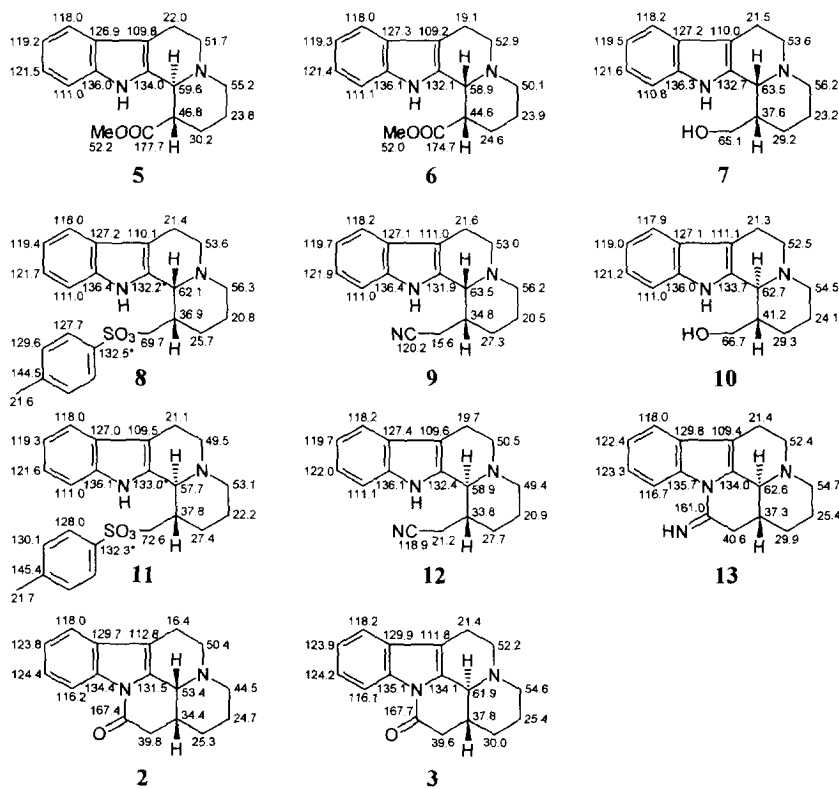
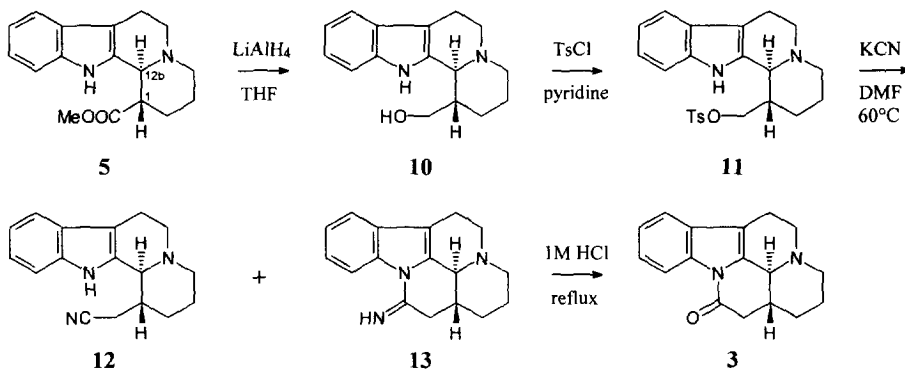


Chart 1

Conclusions

The method described above represents the first selective routes to both deethyleburnamonines at will. Previous reports in the literature include the preparation of deethyleburnamonines from deethylvincamine^{13a}, from methyl (3-pyridyl)acetate^{13b}, from a deethyl derivative of Wenkert's enamine^{13c}, from 3,4-dihydro- β -carboline^{13d}, and from methyl (3-pyridyl)acetate *via* an α -aminonitrile¹⁴ and *via* regioselective cyclization.¹⁵ Recently Stoit and Pandit described a route where two lactam ester intermediates (4-oxo derivatives of **5** and **6**) were prepared *via* folate models.¹²

Experimental

All reactions were carried out under argon. Solvents were distilled over appropriate drying materials before use. Alkaline work-up: addition of aq NaHCO₃, extraction with CH₂Cl₂ (3x), drying of the combined organic layers with Na₂SO₄, and evaporation of the solvent under vacuum. Melting points were determined with a Gallenkamp melting point apparatus and are uncorrected. IR spectra (cm⁻¹, in CHCl₃ unless otherwise noted) were recorded on a Perkin-Elmer 700 spectrophotometer. ¹H NMR (400 MHz, reference: TMS, δ_H = 0.0 ppm) and ¹³C NMR (100 MHz, reference: CDCl₃, δ_C = 77.0 ppm) spectra were recorded on a Varian Unity 400 spectrometer using CDCl₃ as solvent. Coupling constants (*J*) are given in Hz. Signal assignments are based on standard APT, COSY and HETCOR experiments. For the ¹³C NMR data of compounds **2**, **3**, and **5** - **13**, see Chart 1. EI and HR mass spectra (70 eV, *m/z*) were measured with a Jeol DX 303/DA 5000 mass spectrometer. Merck Kieselgel 60 (230-400 mesh) was used in column chromatography.

Ester 5. Prepared according to a literature procedure^{5,6} from tryptophyl bromide and methyl nicotinate in three steps. Yield 67% overall; mp. 144-146°C (Et₂O), lit.⁶ mp. 143-145°C (Et₂O); IR: 2830-2750 (Wenkert-Bohlmann bands), 1715 (C=O); ¹H NMR: 8.15 (1H, br s, NH), 7.48-7.04 (4H, m, arom.), 3.82 (1H, d, *J* = 10, H-12b), 3.80 (3H, s, -COOCH₃); MS: 284 (M⁺, 100), 283 (68), 225 (10), 197 (43), 170 (40), 169 (44); HR-MS: calcd for C₁₇H₂₀N₂O₂: 284.1525, found: 284.1506.

Acid-catalyzed Epimerization of Ester 5 to Ester 6. Ester **5** (197 mg, 0.69 mmol) was dissolved in trifluoroacetic acid (5 ml) and the solution was refluxed overnight (16 h). After evaporation of TFA, alkaline work-up of the residue gave 197 mg of a mixture of esters **5** and **6**. Column chromatography (CH₂Cl₂/MeOH, 99:1) gave 44 mg (22%) of starting ester **5** and 153 mg (78%) of ester **6**; mp. 159-161°C (CH₂Cl₂/hexane); IR: 1715 (C=O); ¹H NMR: 8.41 (1H, br s, NH), 7.48-7.06 (4H, m, arom.), 4.14 (1H, br s, H-12b), 3.65 (3H, s, -COOCH₃); MS: 284 (M⁺, 83), 283 (100), 225 (9), 197 (47), 170 (30), 169 (26); HR-MS: calcd for C₁₇H₂₀N₂O₂: 284.1525, found: 284.1524.

Alcohol 7. Ester **6** (224 mg, 0.79 mmol) in dry THF (20 ml) was added to a suspension of LiAlH₄ (150 mg, 3.95 mmol) in dry THF (20 ml). After 1h stirring, alkaline work-up (aq NaOH) gave 190 mg (94%) of pure alcohol **7**; mp. 195-197°C (MeOH); IR: 3300 (OH), 2830-2750 (Wenkert-Bohlmann bands); ¹H NMR: 7.78 (1H, br s, NH), 7.50-7.07 (4H, m, arom.), 3.85-3.62 (2H, m, -CH₂OH), 3.66 (1H, br s, H-12b); MS: 256 (M⁺, 88), 255 (100), 239 (42), 225 (22), 197 (28), 170 (23), 169 (25); HR-MS: calcd for C₁₆H₂₀N₂O: 256.1576, found: 256.1565.

Tosylate 8. Alcohol **7** (170 mg, 0.66 mmol) was dissolved in dry pyridine (4 ml) and freshly recrystallized *p*-TsCl (253 mg, 1.33 mmol) was added. The mixture was kept at -20°C for 30 h, after which the solvent was evaporated. Alkaline work-up of the residue in the usual manner, followed by column chromatography (CH₂Cl₂/MeOH, 98:2), yielded 177 mg (65%) of amorphous **8**; IR: 3400 (NH); ¹H NMR:

7.91 (1H, br s, NH), 7.49 (2H, d, $J = 8.5$, H-3' and H-5' of Ts), 7.07 (2H, d, $J = 8.5$, H-2' and H-6' of Ts), 7.45-7.05 (4H, m, arom.), 4.21 (1H, dd, $J = 10.5$ and 9, $-\text{CH}_2\text{OTs}$), 3.79 (1H, dd, $J = 10.5$ and 4, $-\text{CH}_2\text{OTs}$), 3.43 (1H, br s, H-12b), 2.34 (3H, s, $-\text{CH}_3$ of Ts); MS: 410 (M^+ , 5), 239 (75), 238 (100), 237 (48); HR-MS: calcd for $\text{C}_{16}\text{H}_{19}\text{N}_2$ (M-OTs) $^+$: 239.1548, found: 239.1546.

Nitrile 9. Tosylate **8** (151 mg, 0.37 mmol) and KCN (133 mg, 2.0 mmol) were heated in DMF (5 ml) at 60°C for 16 h. DMF was evaporated and alkaline work-up of the residue gave the crude product, which was purified by column chromatography ($\text{CH}_2\text{Cl}_2/\text{MeOH}$, 98:2) to give 99 mg (100%) of nitrile **9**; mp. 182-183°C ($\text{CH}_2\text{Cl}_2/\text{hexane}$), lit.⁸ mp. 185-186°C (EtOH); IR: 2830-2750 (Wenkert-Bohlmann bands), 2250 (CN); ^1H NMR: 7.77 (1H, br s, NH), 7.49-7.07 (4H, m, arom.), 3.50 (1H, br d, $J = 2$, H-12b); MS: 265 (M^+ , 76), 264 (100), 239 (9), 225 (11), 197 (62), 170 (74), 169 (58); HR-MS: calcd for $\text{C}_{17}\text{H}_{19}\text{N}_3$: 265.1579, found: 265.1570.

(\pm)-*cis*-Deethyleburnamonine (20-Deethyleburnamonine) (2). Nitrile **9** (26 mg, 0.096 mmol) was refluxed in a 2:1 mixture of 37% HCl and MeOH (3 ml) for 16 h. Alkaline work-up and column chromatography ($\text{CH}_2\text{Cl}_2/\text{MeOH}$, 96:4) gave 21 mg (82%) of (\pm)-*cis*-deethyleburnamonine (**2**); mp. 152-154°C ($\text{CH}_2\text{Cl}_2/\text{hexane}$), lit.^{13c} mp. 156°C (hexane); IR: 1695 (C=O); ^1H NMR: 8.38 (1H, m, H-12), 7.46-7.25 (3H, m, arom.), 4.35 (1H, dt, $J = 5$ and 2.5, H-21); MS: 266 (M^+ , 100), 265 (93), 222 (40), 209 (81), 180 (21), 168 (29), 167 (29); HR-MS: calcd for $\text{C}_{17}\text{H}_{18}\text{N}_2\text{O}$: 266.1419, found: 266.1437.

Alcohol 10. Prepared from ester **5** as described above (see alcohol **7**). Yield 100%; mp. 242-245°C (dec.) (MeOH), lit.¹² mp. 244-245°C (dec.) (MeOH); IR (KBr): 3400 (OH); ^1H NMR: 9.48 (1H, br s, NH), 7.49-7.04 (4H, m, arom.), 4.00 (1H, dd, $J = 11.5$ and 3.5, $-\text{CH}_2\text{OH}$), 3.81 (1H, dd, $J = 11.5$ and 4.5, $-\text{CH}_2\text{OH}$), 3.55 (1H, d, $J = 8.5$, H-12b); MS: 256 (M^+ , 89), 255 (100), 239 (44), 225 (22), 197 (27), 170 (22), 169 (25); HR-MS: calcd for $\text{C}_{16}\text{H}_{20}\text{N}_2\text{O}$: 256.1576, found: 256.1578.

Tosylate 11. Prepared from alcohol **10** as described above (see tosylate **8**). Yield 80%; amorphous; IR: 3400 (NH); ^1H NMR: 8.36 (1H, br s, NH), 7.86 (2H, d, $J = 8.5$, H-3' and H-5' of Ts), 7.39 (2H, d, $J = 8.5$, H-2' and H-6' of Ts), 7.48-7.05 (4H, m, arom.), 4.64 (1H, dd, $J = 10$ and 4, $-\text{CH}_2\text{OTs}$), 4.04 (1H, dd, $J = 10$ and 3, $-\text{CH}_2\text{OTs}$), 3.82 (1H, d, $J = 8$, H-12b), 2.48 (3H, s, $-\text{CH}_3$ of Ts); MS: 410 (M^+ , 23), 239 (87), 238 (100); HR-MS: calcd for $\text{C}_{23}\text{H}_{26}\text{N}_2\text{O}_3\text{S}$: 410.1664, found: 410.1664.

(\pm)-*trans*-Deethyleburnamonine (21-Epi-20-deethyleburnamonine) (3) via Nitrile 12 and Imine 13. Tosylate **11** (82 mg, 0.20 mmol) was treated with KCN (65 mg, 1.0 mmol) in DMF (3 ml) as above (60°C, 20h). The crude mixture, containing mainly nitrile **12** and imine **13** was then refluxed with 1M HCl for 1h. After alkaline work-up of the cooled solution, the crude product was purified by column chromatography ($\text{CH}_2\text{Cl}_2/\text{MeOH}$, 98.5:1.5) to give 30 mg (55% from **11**) of (\pm)-*trans*-deethyleburnamonine (**3**). For obtaining their analytical data, intermediates **12** and **13** were isolated from another experiment. Nitrile **12**: mp. 169-170°C ($\text{CH}_2\text{Cl}_2/\text{hexane}$); IR: 2250 (CN); ^1H NMR: 7.86 (1H, br s, NH), 7.50-7.09 (4H, m, arom.), 3.95 (1H, d, $J = 6$, H-12b); MS: 265 (M^+ , 79), 264 (100), 197 (60), 170 (66), 169 (50); HR-MS: calcd for $\text{C}_{17}\text{H}_{19}\text{N}_3$: 265.1579, found: 265.1564.

Imine **13**: amorphous; IR: 2830-2750 (Wenkert-Bohlmann bands), 1630 (C=N); ^1H NMR: 8.57 (1H, d, $J = 8$, H-12), 7.45-7.19 (3H, m, arom.); MS: 265 (M^+ , 86), 264 (100), 221 (19); HR-MS: calcd for $\text{C}_{17}\text{H}_{19}\text{N}_3$: 265.1579, found: 265.1571.

(\pm)-*trans*-Deethyleburnamonine (3): mp. 145-146°C ($\text{CH}_2\text{Cl}_2/\text{hexane}$), lit.^{13b} mp. 146°C (acetone/hexane); IR: 2830-2750 (Wenkert-Bohlmann bands), 1705 (C=O); ^1H NMR: 8.34 (1H, m, H-12), 7.43-7.22 (3H,

m, arom.); MS: 266 (M^+ , 94), 265 (100), 209 (17), 180 (17), 169 (15), 168 (22), 167 (24); HR-MS: calcd for $C_{17}H_{18}N_2O$: 266.1419, found: 266.1436.

References and Notes

1. Lounasmaa, M.; Tolvanen, A. in *"The Alkaloids"* (Cordell, G. A., Ed.), Vol. 42, Academic Press, New York, 1992, pp. 1-116.
2. Aurousseau, M.; Albert, O.; Huet, Y. in *"Symposium on Pharmacology of Vinca Alkaloids"*, Second Congress of the Hungarian Pharmacological Society (Knoll, J.; Fekete, Gy., Eds.), Vol. 5, Akadémiai Kiadó, Budapest, 1976, pp. 5-8. See also: Vereczkey, L. *Eur. J. Drug Metab. Pharmacokinet.* **1985**, *10*, 89-103.
3. Creasey, W. A., in *"Monoterpenoid Indole Alkaloids"* (Saxton, J. E., Ed.), in *The Chemistry of Heterocyclic Compounds* (Taylor, E. C., Ed.), Supplement to Vol. 25, Part 4, Wiley, Chichester, 1994, pp. 733-736.
4. (a) Lounasmaa, M.; Tolvanen, A. *J. Org. Chem.* **1990**, *55*, 4044-4047; (b) Tolvanen, A.; Din Belle, D.; Lounasmaa, M. *Helv. Chim. Acta* **1994**, *77*, 709-715; (c) Lounasmaa, M.; Din Belle, D.; Tolvanen, A. *Tetrahedron Lett.* **1994**, *35*, 6151-6154; (d) Lounasmaa, M.; Din Belle, D.; Tolvanen, A. *Liebigs Ann.* **1995**, 1385-1387; (e) Lounasmaa, M.; Din Belle, D.; Tolvanen, A. *Tetrahedron Lett.* **1995**, *36*, 7141-7144; (f) Din Belle, D.; Tolvanen, A.; Lounasmaa, M. *Tetrahedron Lett.* **1995**, *36*, 9559-9560; (g) Lounasmaa, M.; Karinen, K.; Din Belle, D.; Tolvanen, A. *Tetrahedron Lett.* **1996**, *37*, 1513-1516.
5. Wenkert, E.; Dave, K. G.; Haglid, F. *J. Am. Chem. Soc.* **1965**, *87*, 5461-5467.
6. Lounasmaa, M.; Johansson, C.-J. *Acta Chem. Scand. B* **1975**, *29*, 655-661.
7. Rosentreter, U.; Born, L.; Kurz, J. *J. Org. Chem.* **1986**, *51*, 1165-1171. See also: (a) Gaskell, A. J.; Joule, J. A. *Tetrahedron* **1967**, *23*, 4053-4063; (b) Zhang, L.-H.; Gupta, A. K.; Cook, J. M. *J. Org. Chem.* **1989**, *54*, 4708-4712.
8. Base-catalyzed cyclization of nitrile **9** to the tautomeric imine fails: Spi, J.; Szab, L.; Thurner, A.; Baitz-Gacs, E.; Tams, J.; Kalaus, Gy.; Szntay, Cs. *Liebigs Ann. Chem.* **1990**, 1133-1136.
9. Lounasmaa, M.; Jokela, R. *Tetrahedron* **1978**, *34*, 1841-1844.
10. For a discussion of these conformations, see ref. 11 and references cited therein.
11. Lounasmaa, M.; Jokela, R.; Tamminen, T. *Heterocycles* **1985**, *23*, 1367-1371. For a reassignment, see ref. 14.
12. Stoit, A. R.; Pandit, U. P. *Tetrahedron* **1989**, *45*, 849-854.
13. (a) Thal, C.; Svenet, T.; Husson, H.-P.; Potier, P. *C. R. Acad. Sci. Ser. C* **1972**, *275*, 1295-1297; (b) Thal, C.; Imbert, T.; Husson, H.-P.; Potier, P. *Bull. Soc. Chim. Fr.* **1973**, 2010-2013; (c) Husson, H.-P.; Imbert, T.; Thal, C.; Potier, P. *Bull. Soc. Chim. Fr.* **1973**, 2013-2016; (d) Imbert, T.; Thal, C.; Husson, H.-P.; Potier, P. *Bull. Soc. Chim. Fr.* **1973**, 2705-2709.
14. Jokela, R.; Karvinen, E.; Tolvanen, A.; Lounasmaa, M. *Tetrahedron* **1988**, *44*, 2367-2375.
15. Massiot, G.; Cherif, A. *Bull. Soc. Chim. Fr.* **1990**, *129*, 648-655.